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The concentration of either the H^+ ion or the OH^- ion is given for four aqueous solutions at 298 K. For each solution, calculate $[H^+]$ or $[OH^-]$. State whether the solution is acidic,

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Matter and Change • Chapter 18 357 Section 18.1 Introduction to Acids and Bases pages 634–643
Practice Problems pages 635–640 1. Write balanced equations for reactions between the following. a.
aluminum and sulfuric acid $2\text{Al}(s) + 3\text{H}_2\text{SO}_4(aq) \rightarrow 2\text{Al}_2(\text{SO}_4)_3$

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